



AIYSEE ENGINEERING SYLLABUS

PHYSICS

1. Physical world and Measurement

- o Units and Dimensions, Errors, Significant Figures
- Measurements and their accuracy

2. Kinematics

o Motion in a Straight Line, Motion in a Plane, Relative Velocity

3. Laws of Motion

o Newton's Laws, Friction, Circular Motion, Free Body Diagrams

4. Work, Energy, and Power

o Work-Energy Theorem, Power, Potential Energy, Conservation of Energy

5. Rotational Motion

o Torque, Angular Momentum, Moment of Inertia, Rolling Motion

6. Gravitation

o Universal Law of Gravitation, Kepler's Laws, Satellite Motion, Escape Velocity

7. Thermodynamics

o Heat, Work, First Law, Specific Heats, Isothermal & Adiabatic Processes

8. Kinetic Theory of Gases

o Pressure of a Gas, RMS Velocity, Degrees of Freedom, Equipartition

9. Oscillations and Waves

 SHM, Simple Pendulum, Damped and Forced Oscillations, Wave Motion, Sound Waves, Doppler Effect

Class 12 Topics

1. Electrostatics

o Coulomb's Law, Electric Field, Potential, Capacitance, Dielectrics

2. Current Electricity

o Ohm's Law, Series & Parallel Circuits, Kirchhoff's Laws, Wheatstone Bridge

3. Magnetic Effects of Current & Magnetism

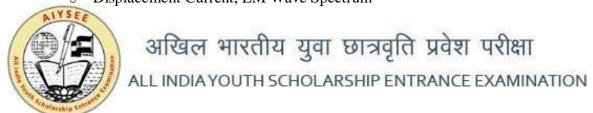
o Biot-Savart Law, Ampere's Law, Lorentz Force, Moving Coil Galvanometer

4. Electromagnetic Induction and Alternating Currents

o Faraday's Laws, Lenz's Law, Eddy Currents, AC Circuits, Resonance

5. Electromagnetic Waves

o Displacement Current, EM Wave Spectrum





- 6. Optics
 - o Reflection, Refraction, Lenses, Wave Optics, Interference, Diffraction, Polarization
- 7. Dual Nature of Matter and Radiation
 - o Photoelectric Effect, de Broglie Waves, Electron Diffraction
- 8. Atoms and Nuclei
 - o Bohr Model, Radioactivity, Nuclear Energy
- 9. Electronic Devices
 - o Semiconductors, Diodes, Transistors, Logic Gates

CHEMISTRY

Class 11 Topics

- 1. Some Basic Concepts of Chemistry
 - o Mole Concept, Stoichiometry, Atomic Mass
- 2. Structure of Atom
 - o Bohr Model, Quantum Numbers, Electronic Configuration
- 3. Classification of Elements and Periodicity
 - o Periodic Table, Periodic Properties
- 4. Chemical Bonding and Molecular Structure
 - o VSEPR Theory, Hybridization, Molecular Orbital Theory
- 5. States of Matter
 - o Gaseous State, Ideal Gas Equation, Kinetic Theory
- 6. Thermodynamics
 - o Enthalpy, Internal Energy, Hess's Law, Spontaneity
- 7. Equilibrium
 - o Le Chatelier's Principle, Ionic Equilibrium, pH
- 8. Redox Reactions
- 9. Hydrogen
- 10. The s-Block Element (Alkali & Alkaline Earth Metals)
- 11. The p-Block Element (Group 13–18 overview)
- 12. Organic Chemistry Some Basic Principles and Techniques
- 13. Hydrocarbons
- 14. Environmental Chemistry



Class 12 Topics

- 1. Solid State
 - o Crystal Lattices, Packing, Defects, Conductivity
- 2. Solutions
 - o Raoult's Law, Colligative Properties
- 3. Electrochemistry
 - o Galvanic Cells, Nernst Equation, Electrolysis
- 4. Chemical Kinetics
 - o Rate Laws, Order of Reaction, Activation Energy
- 5. Surface Chemistry
 - o Adsorption, Catalysis, Colloids
- 6. p-Block Elements (Groups 15-18 in detail)
- 7. d- and f-Block Elements
- 8. Coordination Compounds
- 9. Haloalkanes and Haloarenes
- 10. Alcohols, Phenols, and Ethers
- 11. Aldehydes, Ketones and Carboxylic Acids
- 12. Amines
- 13. Biomolecules
- 14. Polymers
- 15. Chemistry in Everyday Life

MATHEMATICS

Class 11 Topics

- 1. Sets, Relations and Functions
- 2. Complex Numbers and Quadratic Equations
- 3. Permutations and Combinations
- 4. Binomial Theorem
- 5. Sequence and Series
- 6. Straight Lines and Conic Sections (Parabola, Ellipse, Hyperbola)
- 7. Limits and Derivatives (Introduction)
- 8. Mathematical Induction
- 9. Statistics and Probability
- 10. **Trigonometry**





Class 12 Topics

- 1. Relations and Functions (Inverse Trig Functions)
- 2. Matrices and Determinants
- 3. Continuity, Differentiability and Applications of Derivatives
- 4. Integrals and Applications of Integrals
- 5. Differential Equations
- 6. Vectors
- 7. Three-Dimensional Geometry
- 8. Probability (Advanced)
- 9. Linear Programming